

Plasma catalysis in liquid water for CO₂ conversion

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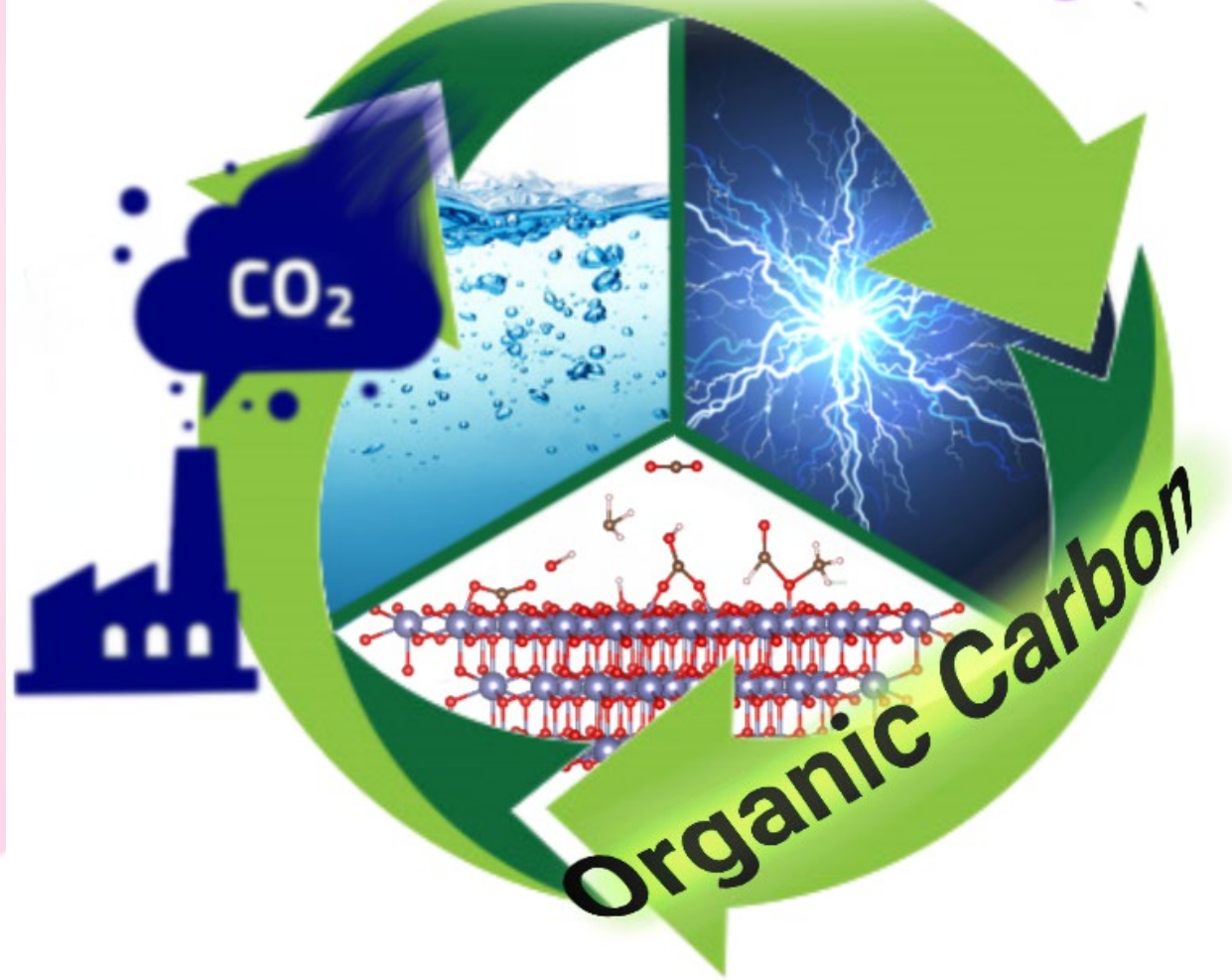
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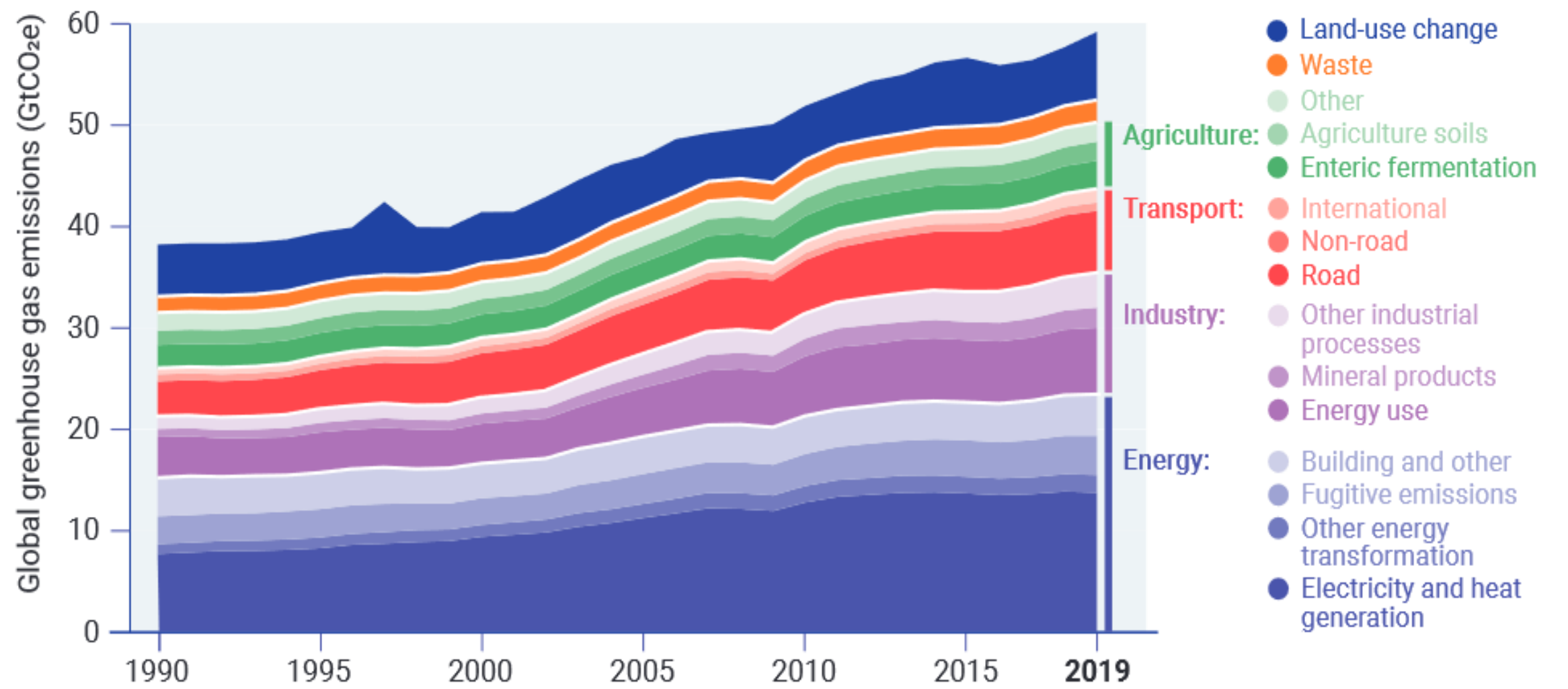
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Plasma Catalysis



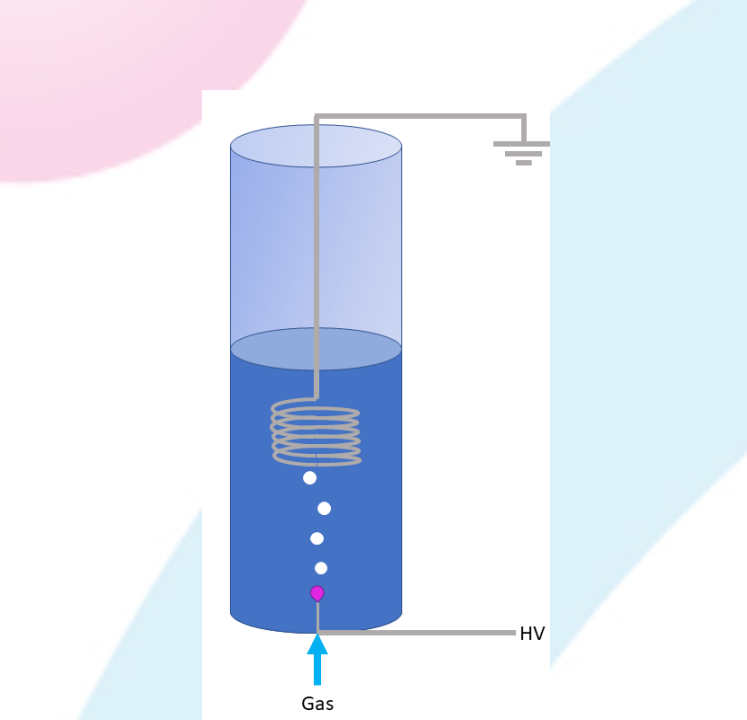
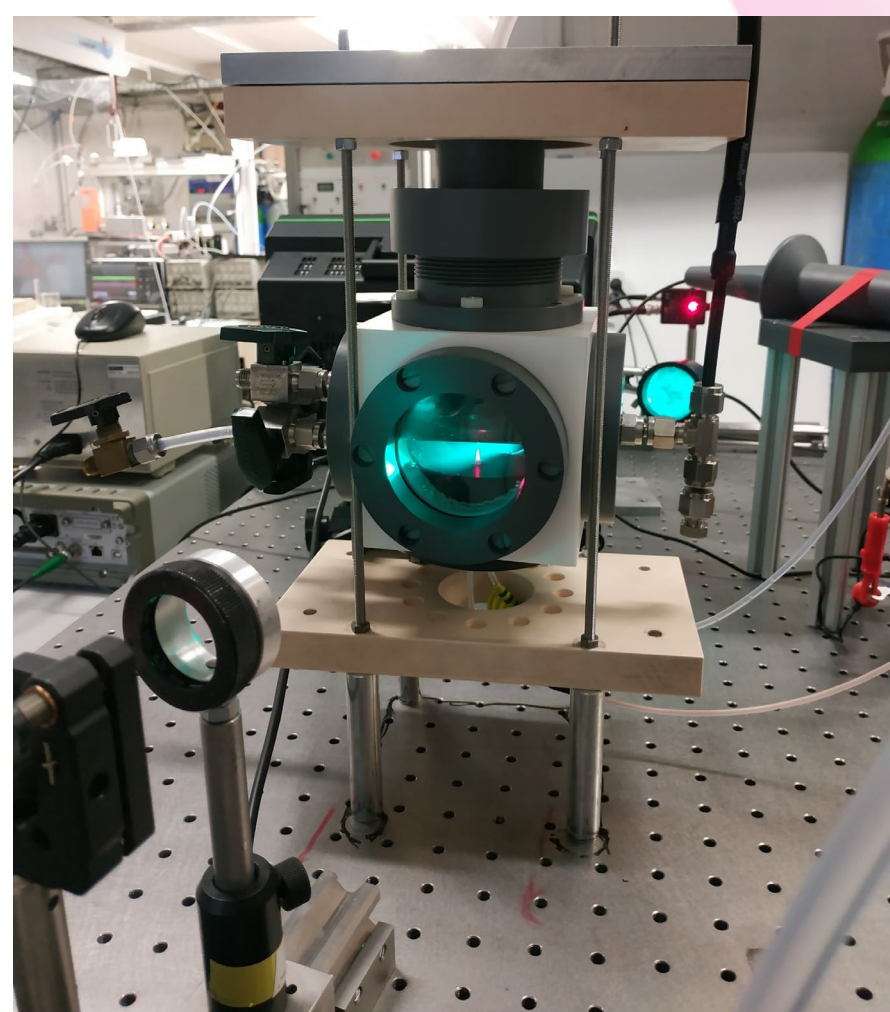
GHG emissions at the sectoral level.



Source: Crippa et al. (2020)

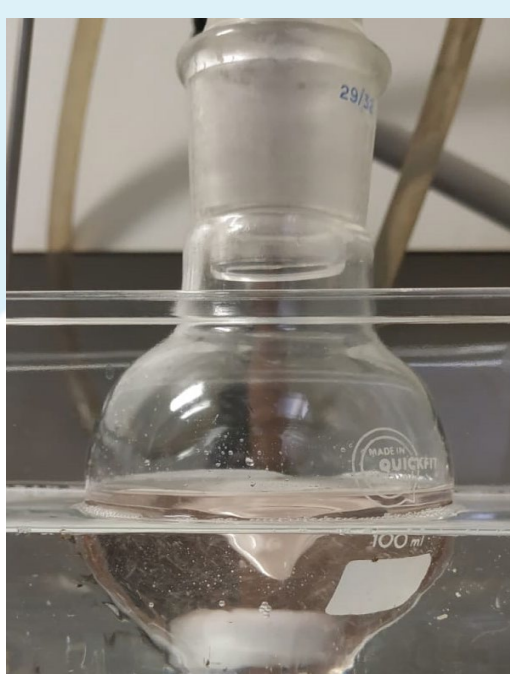
United Nations Environment Programme (2020). *Emissions Gap Report 2020*. Nairobi

Non-thermal Liquid phase plasma-catalytic reactor



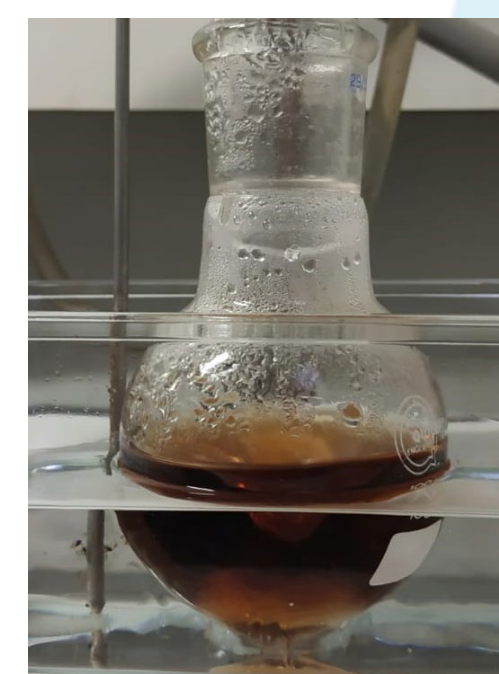
- Catalyst suspended in the liquid
- 20 kV, 300 Hz, bubbling CO₂
- Laser triggering power supply
- High speed camera

Glycerol + Mn(NO₃)₂



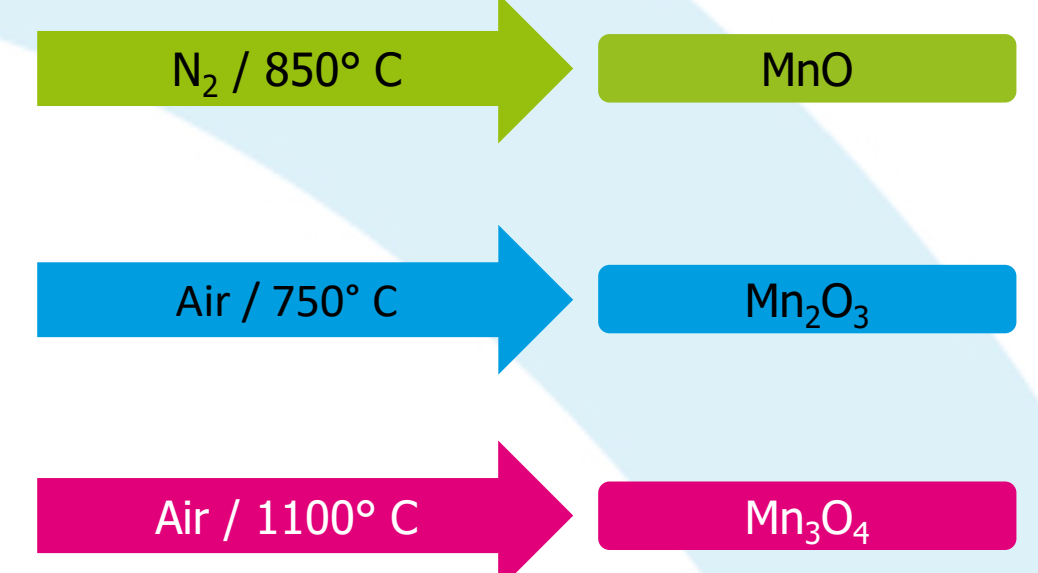
180° C

Manganese(II) Glycerolate



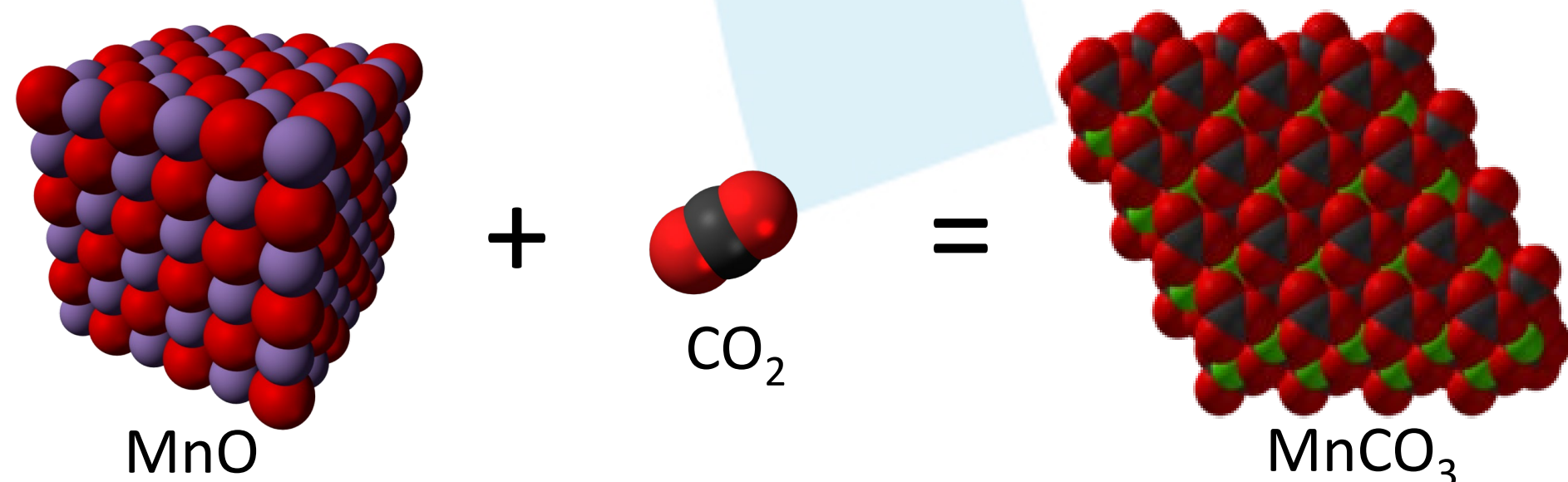
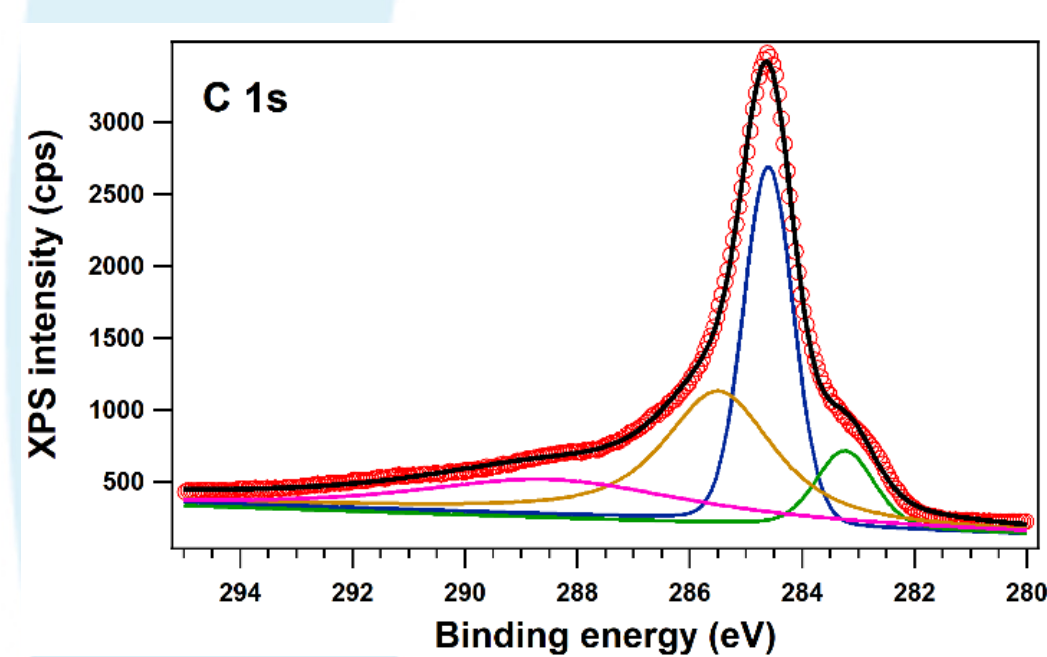
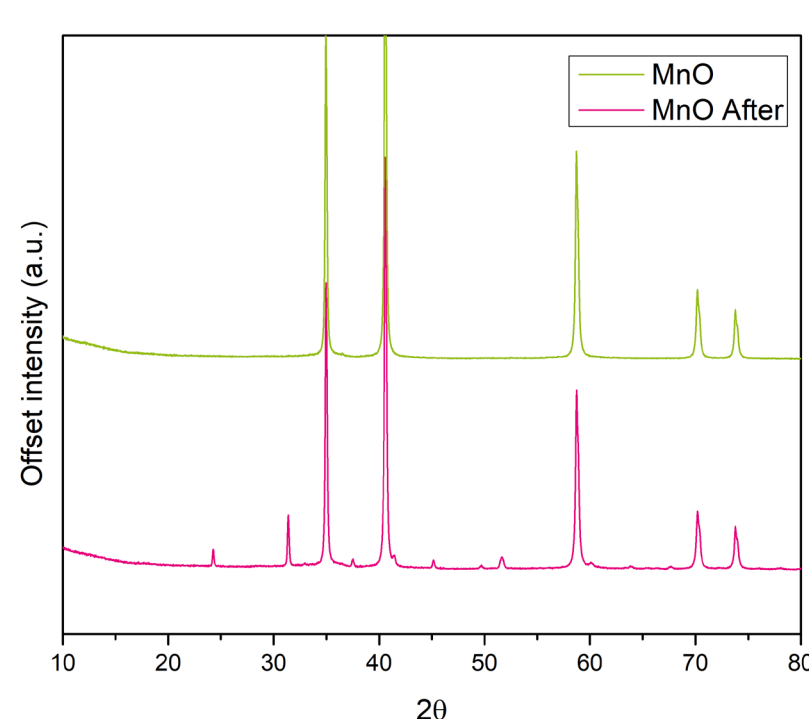
- Green synthesis of MnO, Mn₂O₃ and Mn₃O₄
- Different oxidation states allow to investigate catalyst effect and stability.

Synthesis



- Catalyst characterization by XRD and XPS.
- Products analyzed by TOF-MS, TOC and FTIR

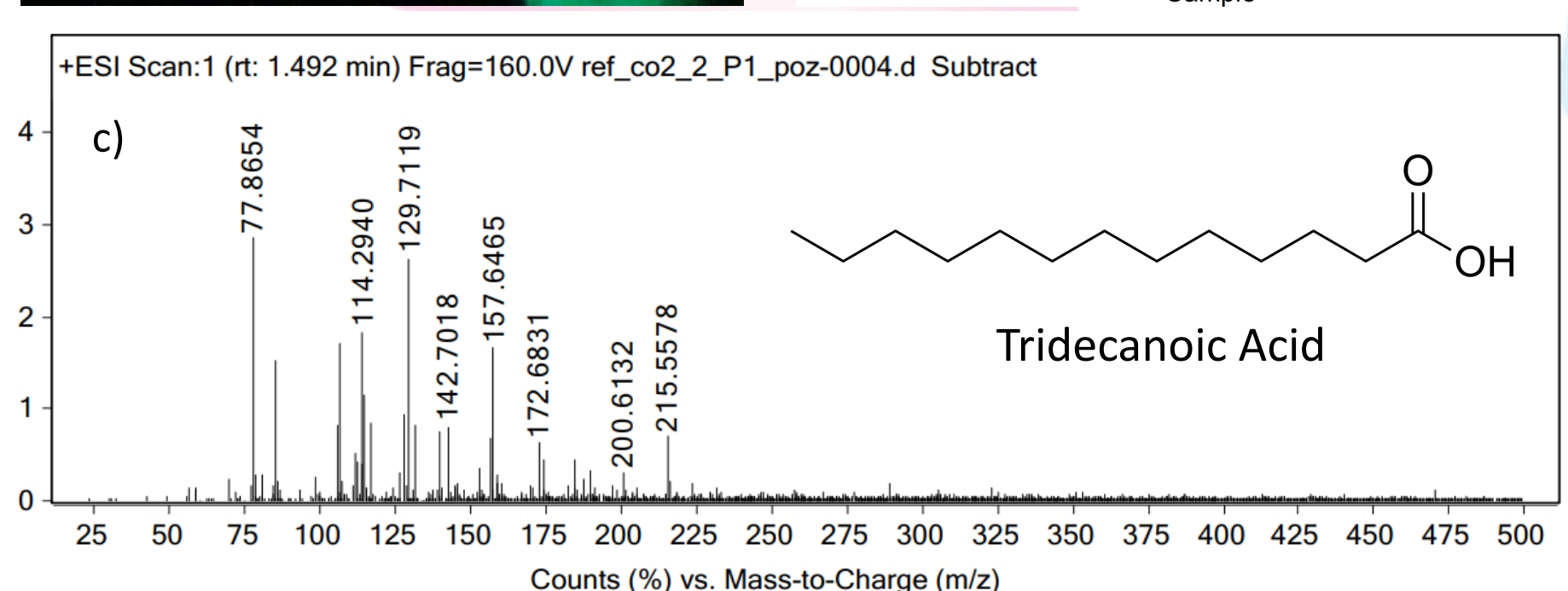
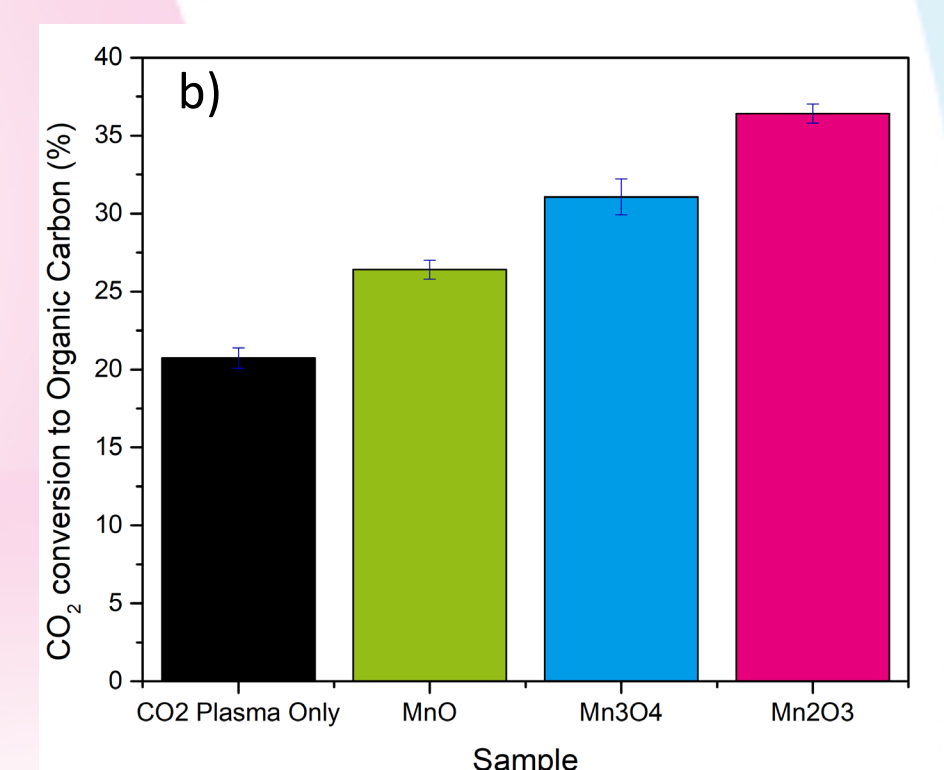
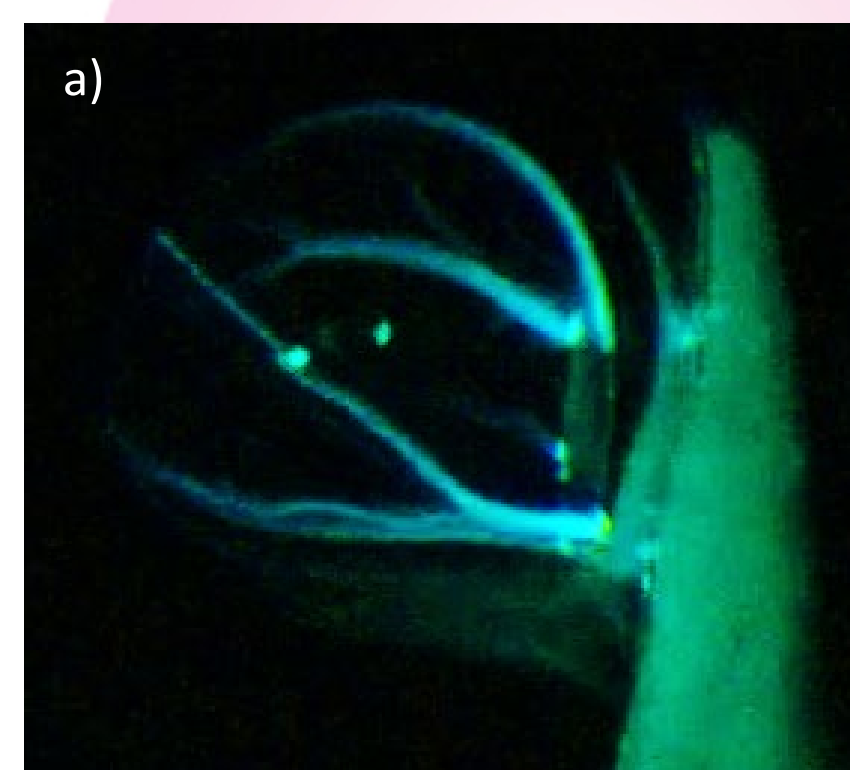
Plasma liquid: new route for CO₂ fixation into solids?



a) XRD MnO before and after plasma reactions. b) XPS MnO after reactions.

- XRD shows the presence of peaks of MnCO₃ after the discharge.
- XPS shows band associated with carbonates.
- MnO did not perform well as a catalyst but it was successfully converted to MnCO₃.
- Fixating CO₂ into solids is also a topic of great research interest and there are still no reports of it being done using plasma and liquid water.

And organic liquids too?



a) Plasma filaments inside CO₂ bubbles. b) Total organic carbon, liquid samples after 15 min of reaction. c) TOF-MS sample CO₂ plasma alone.

- Using liquid water in contact with CO₂ plasma unlocks reaction paths that lead to the formation of long-chain organic acids.
- More stable catalysts (Mn₂O₃ and Mn₃O₄) performed significantly better than plasma alone.
- Such large molecules can be used to store carbon efficiently, since liquids are more easily stored and transported.
- Being performed in liquid-phase, it can be coupled with other conventional techniques (photo and electrocatalysis), using the same catalysts.