# Water you waiting for? - A Complete and Consistent Set of Electron-H<sub>2</sub>O Collision Cross Sections for Plasma Modelling

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### 1. Introduction

Water is an omnipresent impurity in electric discharges where  $H_2O$  molecules interact with free electrons e<sup>-</sup>. The characterisation of the e<sup>-</sup> kinetics by the electron energy distribution function (EEDF) requires a complete and consistent set of electron collision cross sections (CSs). By the e<sup>-</sup> swarm method, such a set is derived that considers the anisotropy of the rotational scattering through individual differential, integral and momentum transfer cross sections (DCSs, ICSs and MTCSs, respectively) based on the Born approximation [1, 2]. Assuming isotropic scattering simplifies the proposed set, allows for its application in freely accessible space-homogeneous codes and still yields better agreement with experiments compared to other established CS sets.

$$DCS_{i}(\varepsilon,\theta) = \sqrt{\frac{\varepsilon'}{\varepsilon}} \frac{4D^{2}}{6(2J'+1)} \frac{S_{i}}{\varepsilon'+\varepsilon - 2\sqrt{\varepsilon'\varepsilon}\cos\theta} \quad \text{such that} \quad ICS_{i}(\varepsilon) = 2\pi \int_{0}^{\pi} DCS_{i}(\varepsilon,\theta)\sin\theta d\theta \quad \text{and} \quad MTCS_{i}(\varepsilon) = 2\pi \int_{0}^{\pi} DCS_{i}(\varepsilon,\theta)(1-\cos\theta)\sin\theta d\theta$$



#### 3. Validation



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Fig. 1: Proposed complete and consistent electron collision cross section set for water molecules divided in conservative (number of electrons is constant) and non-conservative (ionisation and attachment) collisions. When assuming isotropy, the rotational ICSs and MTCSs are adjusted or discarded, respectively [3].



Fig. 2: Comparison of experimental electron swarm parameters - from top

Fig. 3: Agreement of two-term solvers LoKI-B and BOLSIG+ [5] as well as Monte Carlo code LoKI-MC [6] shown by means of the red. mobility (left), EEDFs at representative E/N with and without included rotations (right).

#### 5. Conclusion

- Proposition of complete and consistent set of electron CSs for  $H_2O$
- Simplification by assuming isotropy allows for wide-spread applicability
- Inclusion of anisotropy improves results (even beyond two-term solvers)

• Release of both sets in the IST-LISBON database on LXCat [7]

to bottom: drift velocity, red. mobility, char. energy, red. Townsend coef., red. eff. Townsend coef. and red. attachment coef. - with LoKI-B [4] results either assuming isotropy or including anisotropy vs. the red. el. field E/N.

## 6. Literature

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