Study of CO₂-CH₄ plasma-surface interactions on cerium oxide using in situ FTIR transmission experiments

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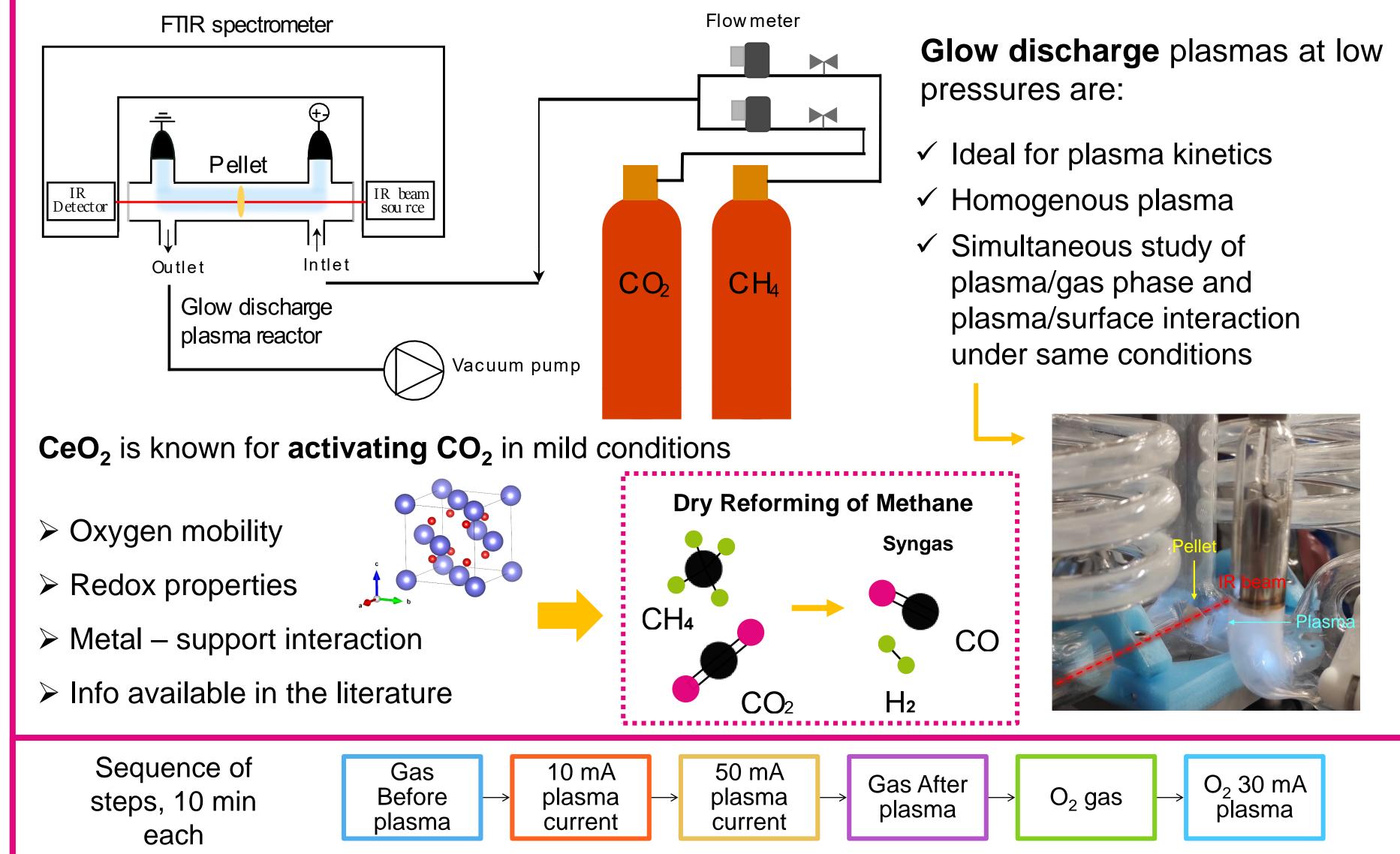
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Using Plasma-Catalysis for CO₂ recycling

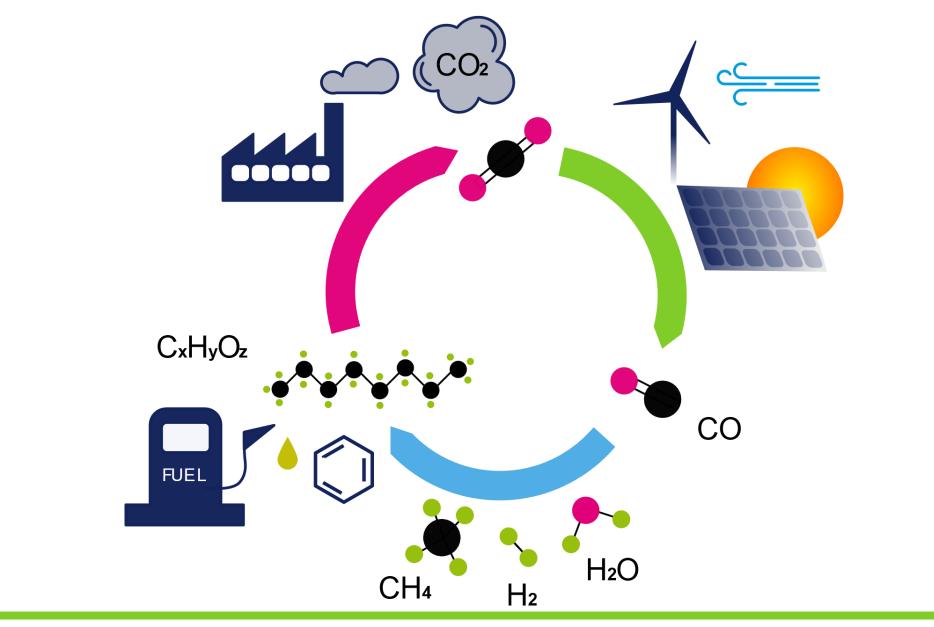
Climate change caused by greenhouse gas emissions encourage the closing of the carbon cycle which can be achieved by converting CO₂ into platform molecules or short hydrocarbons. The use of **non-thermal plasmas** can promote the asymmetrical stretching vibration of CO for bond breakage. The presence of a catalyst could greatly improve the conversion and selectivity. The complexity of the interaction of plasma with a surface brings the necessity to study the underlying mechanisms occurring on the catalyst as a function of time and under different conditions.^[1,2]

in situ FTIR transmission to study DC glow discharge plasmas

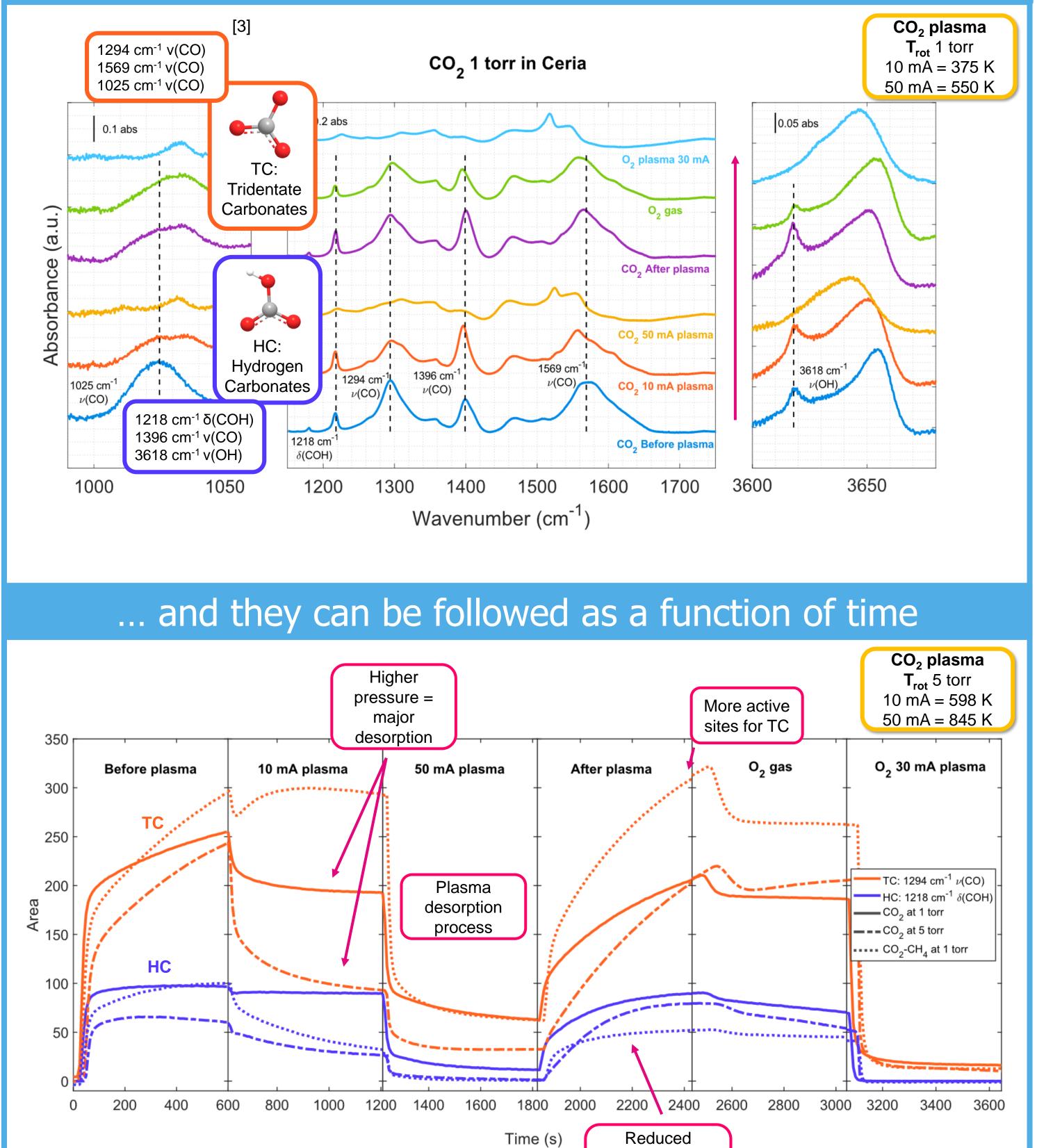




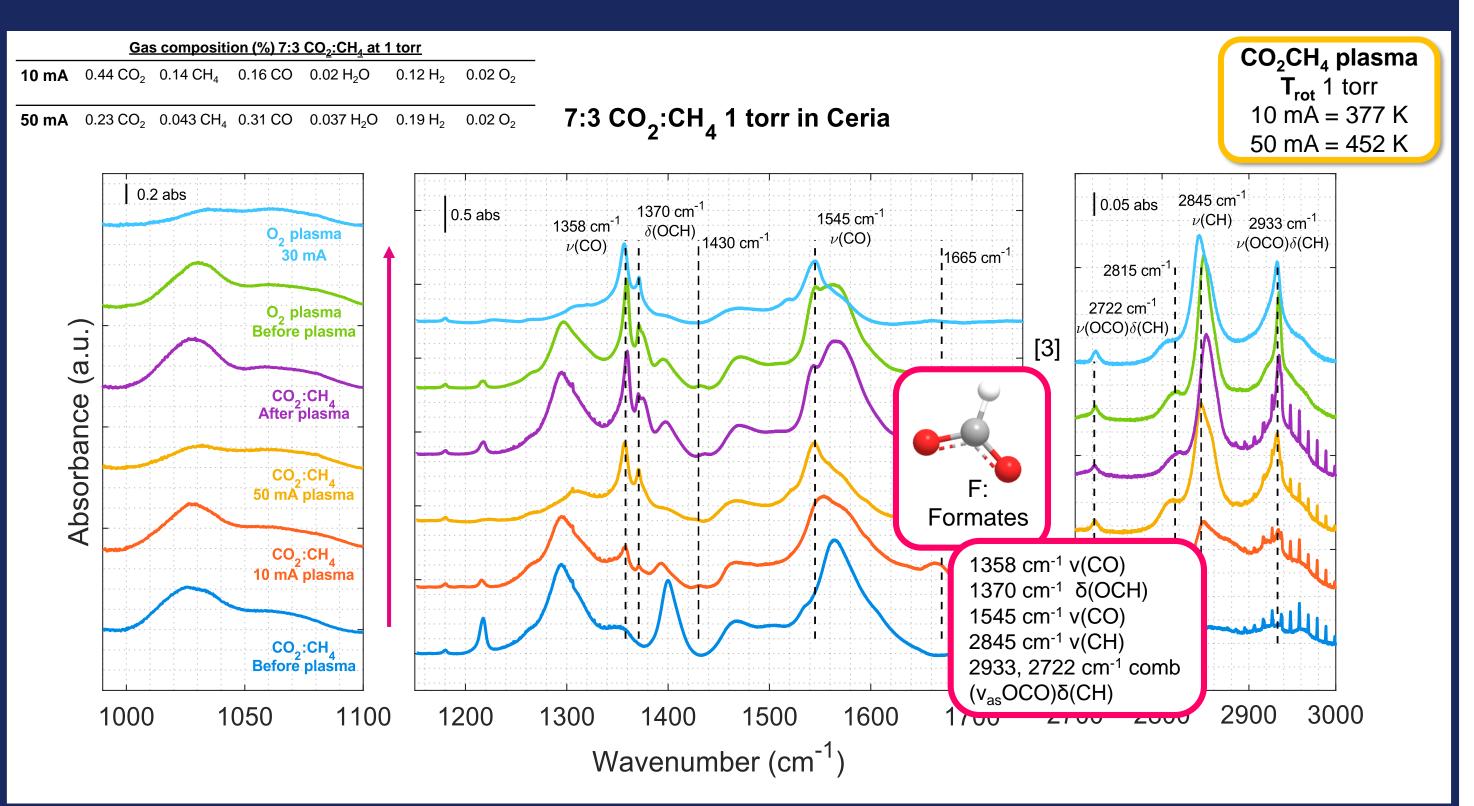
Laboratoire de Physique des Plasmas



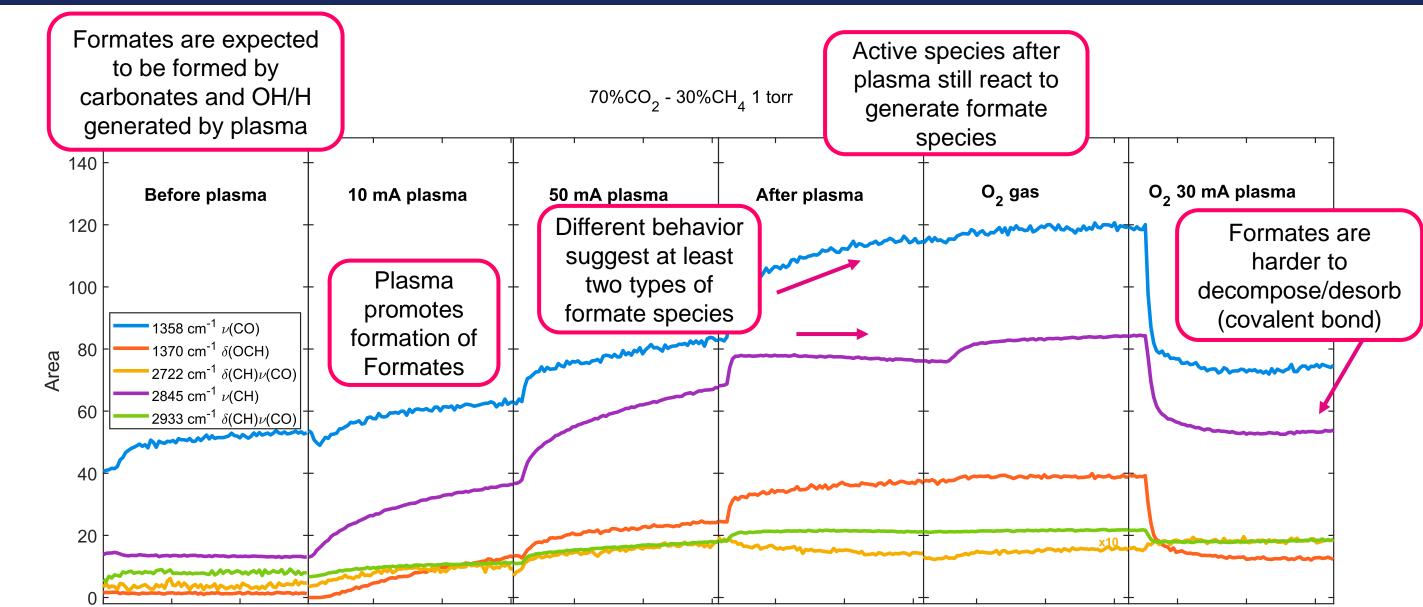
Carbonates are the main species by CO₂ adsorption...

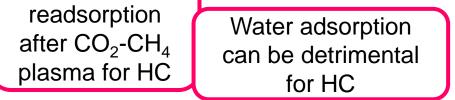


CO₂-CH₄ plasmas promote formate species on the surface of CeO_2



Formates are known as surface intermediates for CO and H₂ release





The addition of CH_4 to the mixture can change the adsorption of carbonates

1800 2000 2200 2400 2600 2800 3000 Time (s)

TC and OH groups are the precursors of Formates species

 $CO_{3(ads)}^{2-} + OH_{(ads)}^{-} \leftrightarrow HCO_{2(ads)}^{-} + O_{2(ads)} \leftrightarrow CO_{(ads)} + OH_{(ads)}^{-}$ Tridentate From dissociated Formate carbonate

References:

[1] R. Vakili 2020 Appl. Catal. B Environ. 260, pp. 118195 [2] Z. Sheng, H. H. Kim, S. Yao, and T. Nozaki, 2020 Phys. Chem. Chem. Phys., 22, pp. 19350 [3] G. N. Vayssilov 2011 J. Phys. Chem. C, 115 pp. 23435-23454

[4] C. Li, 1989 J. Chem. Soc. Faraday Trans. 1 85 pp. 1451–1461 [5] Ferreira-Aparicio et al. 2000, Appl. Catal. A Gen. 202, 183–196

This project has received funding from the European Union's Horizon 2020 research and innovation programme under the Marie Skłodowska-Curie grant agreement No 813393



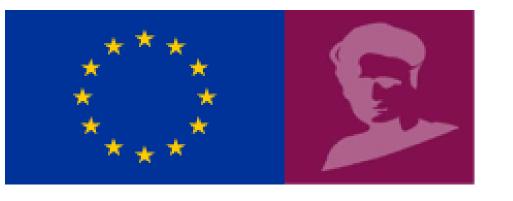




Conclusions

- Tridentate carbonates and OH species are found as precursors for formates species in the ceria surface during CO₂-CH₄ plasma interaction
- Hydrogen carbonates are sensitive to excess H₂O molecules
- Time evolution of the bands suggests at least two types of formate species





[4]

[5]