

# Pretty exciting - Calibrated OH laser-induced fluorescence spectroscopy, opportunities and challenges in unravelling the role of H<sub>2</sub>O in CO<sub>2</sub> plasma conversion

Maik Budde<sup>1,2\*</sup>, Luca Matteo Martini<sup>3</sup>, Matteo Ceppelli<sup>3</sup>, Sara Quercetti<sup>3</sup>, and Richard Engeln<sup>1,4</sup>

<sup>1</sup>Eindhoven University of Technology, Eindhoven, The Netherlands, <sup>2</sup>Instituto de Plasmas e Fusão Nuclear, Instituto Superior Técnico, Universidade de Lisboa, Lisbon, Portugal, <sup>3</sup>Dipartimento di Fisica Università di Trento, Trento, Italy, <sup>4</sup>ASML Netherlands B.V., The Netherlands

\* m.budde@tue.nl

## Introduction

The consideration of water in the plasma conversion of carbon dioxide is crucial due to its role as omnipresent impurity and/or abundant hydrogen source [1]. Quantitative in situ measurements by laser-induced fluorescence (LIF) spectroscopy facilitate our understanding since OH radicals serve here not only as reactant but also as probe of the CO<sub>2</sub> conversion [2, 3]. A CO<sub>2</sub>-H<sub>2</sub>O glow discharge serves as non-equilibrium environment. Quantitative measurements are impeded by spectral overlap of OH with excited CO [4].

## Setup

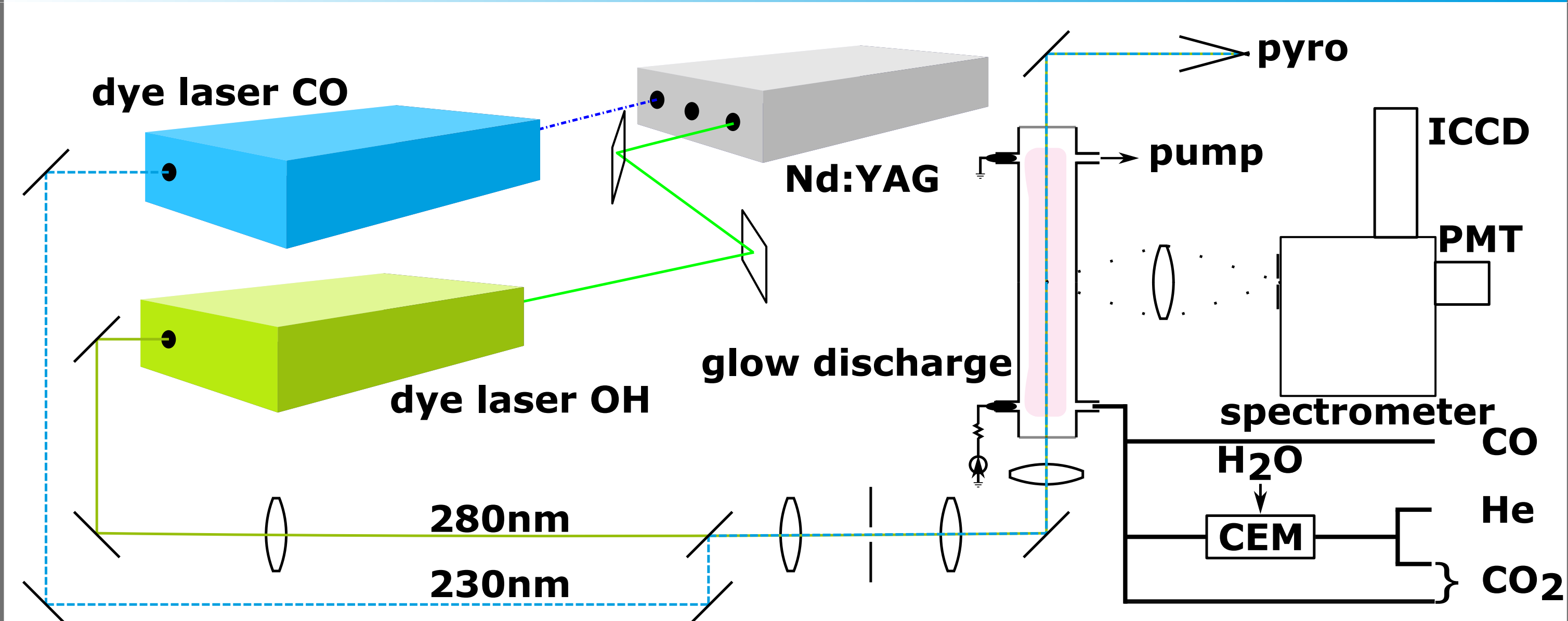


Fig. 1: Used setup for absolute OH LIF measurements in a CO<sub>2</sub>-H<sub>2</sub>O glow discharge at 6.67mbar. Calibration by CO TALIF.

## Spectral Overlap with CO

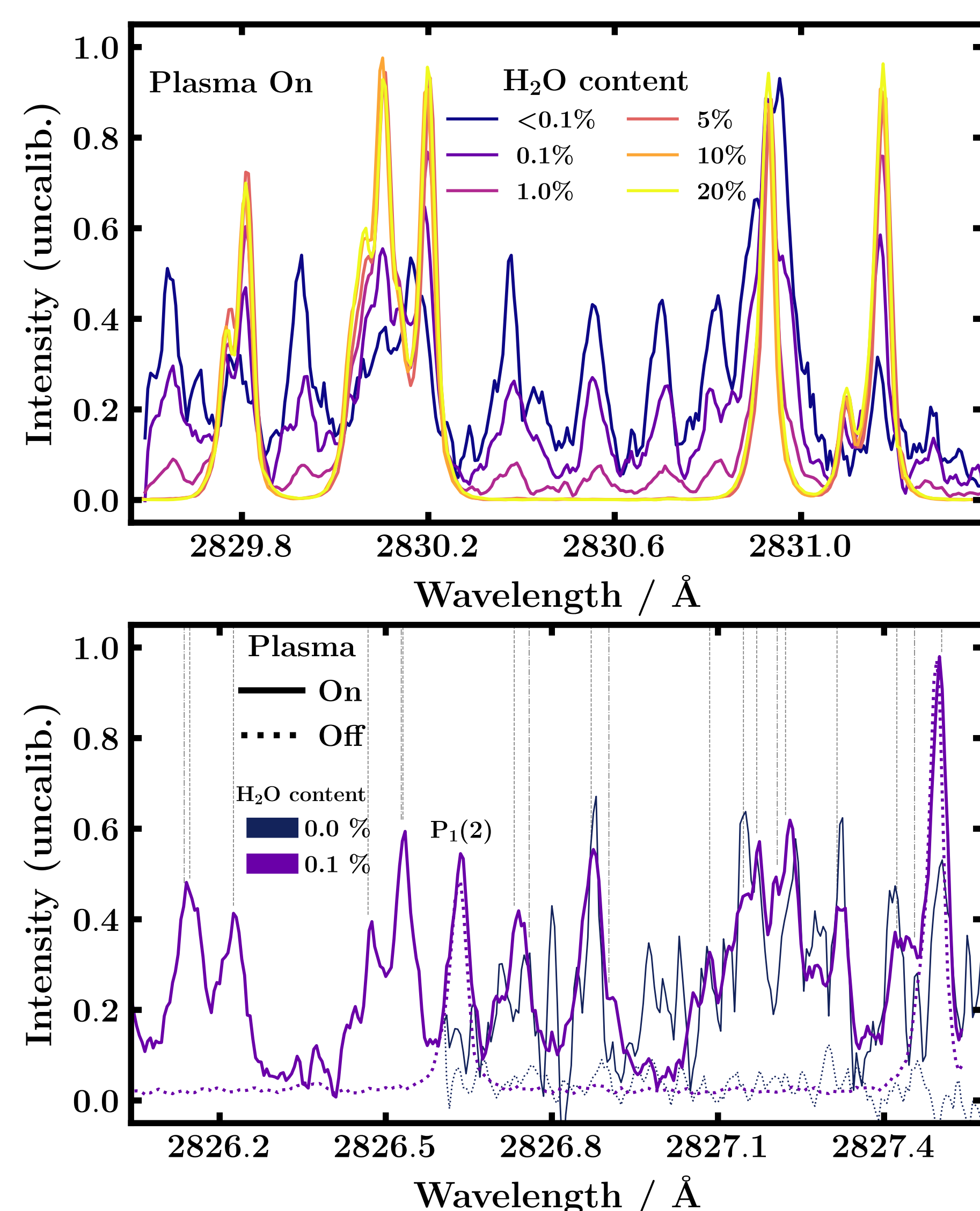


Fig. 3: Distortion of the OH excitation spectrum for decreasing water admixture, due to spectral overlap with the third positive system of CO (top) and proposition of the overlap-free P<sub>1</sub>(2) for quantitative measurements (bottom) [5, 6].

## Literature

1. R. Snoeckx *et al.*, *ChemSusChem* **10**, 409–424 (2017).
2. M. Budde *et al.*, *Plasma Sources Sci. Technol.*, *Accept. Manuscript* (2022).
3. M. Ceppelli *et al.*, *Plasma Sources Sci. Technol.* **29**, 065019 (2020).
4. M. Budde *et al.*, *Appl. Spectrosc.* **0**, 00037028221088591 (2022).
5. R. K. Asundi *et al.*, *Proc. R. Soc. London, Ser. A* **124**, 277–296 (1929).
6. G. H. Dieke *et al.*, *Phys. Rev.* **43**, 12–30 (1933).

## Quantitative LIF in a Glow Discharge

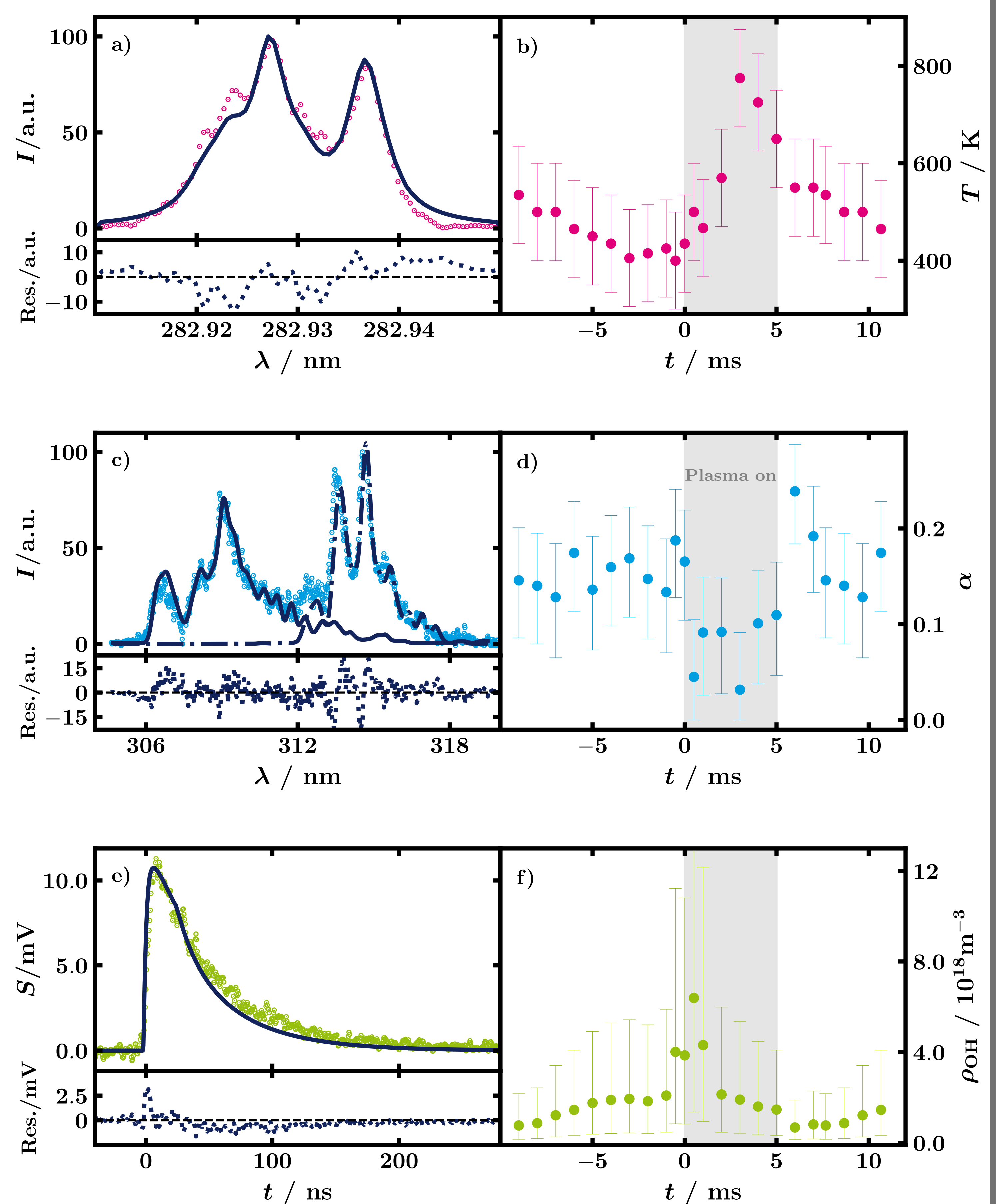
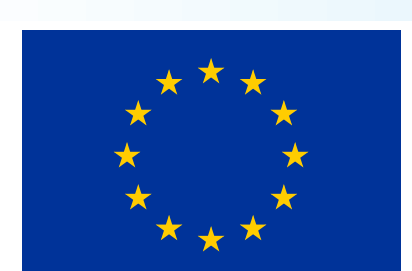


Fig. 2: Simulation of experimental observables like excitation spectrum (a), fluorescence spectrum (c) and fluorescence pulse (e) to obtain the time-dependent rot. temperature  $T$  (b), CO<sub>2</sub> conversion  $\alpha$  (d) and OH density  $\rho_{OH}$  (f) with respect to the plasma.

## Conclusion

- Suggestion of new OH LIF calibration method by CO TALIF
- Measurement of absolute OH number density in pulsed DC CO<sub>2</sub>-H<sub>2</sub>O glow discharge for the first time
- Confirmation of conversion drop in plasma on-time
- Proposition of excitation of P<sub>1</sub>(2) of OH to avoid spectral crosstalk with the third positive system of CO
- Demonstration of single-photon CO LIF

## Funding



This project has received funding from the European Union's Horizon 2020 research and innovation programme under the Marie Skłodowska-Curie grant agreement No 813393